

# Auxiliary strategies for the preparation of $\beta$ -amino alcohols with reductive cross-coupling and a synthesis of (–)-cytoxazone

Xiangjie Lin, Paul A. Bentley\* and Hexin Xie

Department of Chemistry, MSC03 2060, University of New Mexico, Albuquerque, NM 87131-0001, USA

Received 16 August 2005; revised 1 September 2005; accepted 1 September 2005

**Abstract**—Imine auxiliaries including chiral *N*-*tert*-butanesulfinyl imines have been successfully utilized to provide stereochemical control to the reductive cross-coupling of imines with aldehydes or ketones. This methodology has been applied to the synthesis of (–)-cytoxazone.

© 2005 Published by Elsevier Ltd.

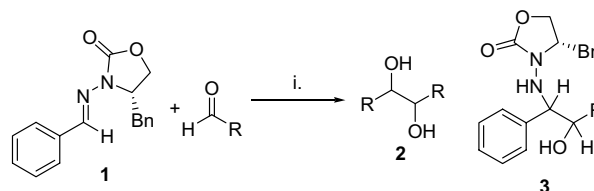
The application of auxiliaries has successfully controlled the stereochemical outcomes in a range of radical reactions.<sup>1</sup> We considered this to be a potential approach towards the preparation of  $\beta$ -amino alcohols from the reductive coupling of imines and carbonyls.<sup>2</sup>  $\beta$ -Amino alcohols are a frequent functional motif in natural products and drugs of significant biological activity<sup>3–5</sup> and have thus merited much synthetic attention.<sup>6–9</sup> However, few approaches offered potential access to  $\beta$ -amino alcohols with high stereoselectivity for either diastereoisomer as secondary or tertiary alcohols and incorporating a carbon–carbon bond forming event. This potential is present in the ‘Pinacol-type’ cross-coupling discussed in this paper.

This ideal picture has been fraught with difficulties of chemo- and diastereo- control.<sup>2</sup> There are a very limited number of examples encompassing enantio-control. Uemura et al. have utilized enantiopure ferrocenyl<sup>10,11</sup> and Cr(CO)<sub>3</sub> aromatic aldehydes,<sup>12</sup> thus prohibiting a wider range of aldehydes for coupling such as aliphatic aldehydes. We viewed the introduction of the chiral auxiliary on the imine nitrogen as offering a more flexible strategy and there existed numerous enantiopure amine precursors. During preparation of this manuscript a similar plan was achieved by Xu and Lin et al.<sup>13</sup>

Identification of an enantiopure auxiliary that would provide high stereocontrol for the cross-coupling reaction was crucial. Initially oxazolidinones (Scheme 1) (e.g., **1**) were focused upon in light of the success that Friestad et al. have found in radical alkylations.<sup>14</sup>

Aromatic aldehydes (e.g., benzaldehyde) were found to be too reactive and only gave a homocoupled product (Table 1; entry 1). However aliphatic aldehydes (Scheme 1) provided more promise and gave moderate to excellent diastereoselectivity, but problematically in poor yield (Table 1; entries 2 and 3).

Amongst other auxiliaries studied were chiral *N*-*tert*-butanesulfinyl imines (**4**) first used by Ellman et al. with great effect in alkylation.<sup>15</sup> Recently, Xu et al. have, utilized this as an auxiliary with high stereocontrol in homocoupling<sup>16</sup> and cross-coupling with nitrones.<sup>17</sup> Thus this auxiliary was selected for cross-coupling initially with aldehydes (Scheme 2).



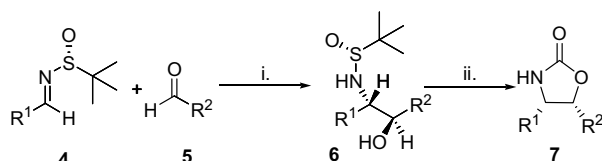
**Scheme 1.** The synthesis of a  $\beta$ -amino alcohol by the cross-coupling of aldehydes with an imine oxazolidinone auxiliary. Reagents and conditions: i. SmI<sub>2</sub>, *t*-BuOH, THF, ZnX<sub>2</sub> (see Table 1), 25 °C.

**Keywords:** Reductive cross-coupling; *N*-*tert*-Butanesulfinyl imine; Chiral auxiliary; Cytosazone; Pinacol.

\*Corresponding author. Tel.: +1 505 243 8073; fax: +1 505 277 2609; e-mail: [pbentley@unm.edu](mailto:pbentley@unm.edu)

**Table 1.** Cross-coupling of aldehydes with an imine oxazolidinone auxiliary shown in Scheme 1

Entry	Reagent/product	Diastereomeric ratio (dr) <sup>a</sup>	Yield (%)
1	X = Br, <b>2</b> R = Ph	—	90
2	X = Br, <b>3</b> R = <sup>n</sup> Bu	20:1	30
3	X = Cl, <b>3</b> R = <sup>i</sup> Pr	5:1	46

<sup>a</sup> Diastereomeric ratio determined by <sup>1</sup>H NMR.**Scheme 2.** The synthesis of a  $\beta$ -amino alcohol by the cross-coupling of aldehydes with an imine (*R*)-*N*-*tert*-butanesulfinyl auxiliary. Reagents and conditions: i.  $\text{SmI}_2$ , <sup>t</sup>BuOH, THF,  $-78^\circ\text{C}$ ; ii. (a) HCl, MeOH,  $25^\circ\text{C}$  (b) triphosgene,  $\text{Et}_3\text{N}$ , DCM,  $25^\circ\text{C}$ .

Given the experiences with the oxazolidinone auxiliary (**1**) aromatic aldehydes were avoided. The cross-coupling in Table 2; entry 2 was studied, initially at  $-20^\circ\text{C}$  giving only 21% yield of the  $\beta$ -amino alcohol (**6**  $\text{R}^1 = \text{Ph}$ ,  $\text{R}^2 = ^i\text{Pr}$ ). However, reduction of the reaction temperature to  $-78^\circ\text{C}$  saw a dramatic improvement. The reaction displayed very good diastereoselectivity (14:1) and chromatography allowed isolation of a single diastereomer (**6**  $\text{R}^1 = \text{Ph}$ ,  $\text{R}^2 = ^i\text{Pr}$ ) in high yield (91%) (Scheme 2).

Following this success, the phenyl imine (**4**  $\text{R}^1 = \text{Ph}$ ) was cross-coupled with a variety of aliphatic aldehydes. Trends in the reactions of **5**, from  $\text{R}^2 = \text{Et}$  to  $\text{R}^2 = ^i\text{Pr}$  to  $\text{R}^2 = ^t\text{Bu}$  (Table 2; entries 1–2–3) suggested that an increase in size of the substituent in close proximity of the aldehyde's  $\alpha$ -carbon displayed increased diastereoselectivity. Thus the aldehyde with the bulkiest substituent (**5**  $\text{R}^2 = ^t\text{Bu}$ ) provided the highest diastereoselectivity, but regrettably at the cost of a low yield and a comparatively long reaction time (Table 2; entry 3). Increase in the size of the aldehyde further from the  $\alpha$ -carbon gave a slight reduction in diastereoselectivity

(e.g., Table 2; entry 1 compared to entry 4, or entry 2 compared to entry 6). Pentanal (**5**  $\text{R}^2 = ^n\text{Bu}$ ) was also found to give good diastereoselectivity (Table 2; entry 5) when coupled with phenyl imine (**4**  $\text{R}^1 = \text{Ph}$ ).

Variation of the aromatic (*R*)-*N*-*tert*-butanesulfinyl imine (**4**) was sought in an attempt to improve the diastereoselectivity of the cross-coupling even further. The poorest aldehyde (**5**  $\text{R}^2 = ^n\text{Bu}$ ), in terms of the cross-coupling's diastereoselectivity (Table 2; entry 4), was the partner for this study. The introduction of electron-withdrawing substituents on the phenyl ring such as *p*-chloro (Table 2; entry 8) caused a slight decrease in the diastereoselectivity and yield of the cross-coupling reaction. However, electron-donating substituents on the phenyl ring such as *p*-methoxy (Table 2; entry 9) provided a significant improvement in the diastereoselectivity of  $\beta$ -amino alcohol formation.

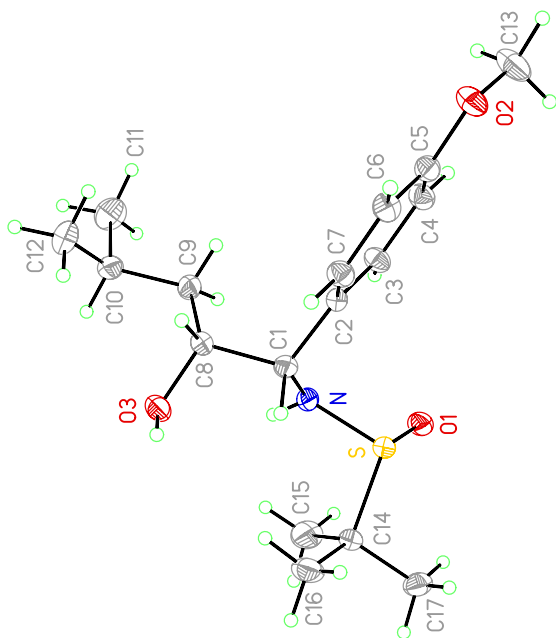
An aliphatic (*R*)-*N*-*tert*-butanesulfinyl imine was also prepared (**4**  $\text{R}^2 = ^i\text{Pr}$ ) and was coupled with **5** ( $\text{R}^2 = ^i\text{Pr}$ ) with good diastereoselectivity (Table 2; entry 11). The (*R*)-*N*-*tert*-butanesulfinyl auxiliary was removed under acidic conditions and the resulting amino alcohol was protected as the oxazolidinone **7** ( $\text{R}^1 = \text{R}^2 = ^i\text{Pr}$ ) in high yield. The *cis*-diastereomer had been previously synthesized,<sup>18</sup> and its  $\text{H}^1$  and  $\text{C}^{13}$  NMR data matched **7** ( $\text{R}^1 = \text{R}^2 = ^i\text{Pr}$ ). Crystallization of **6** ( $\text{R}^1 = p\text{-MeOC}_6\text{H}_4$ ,  $\text{R}^2 = ^t\text{Bu}$ ) from ether allowed investigation by X-ray analysis (Fig. 1) and demonstrated the *anti*-diastereomer to be the major diastereomer.

A derivative of aldehyde **5** ( $\text{R}^2 = \text{CH}_2\text{OBn}$ ), which contained a protected alcohol was perceived to have greater synthetic utility and cross-coupling with **4** ( $\text{R}^1 = \text{Ph}$ ) gave good diastereoselectivity and yield (Table 2; entry 7). Improved diastereoselectivity of the cross-coupling was again observed when the *p*-methoxy substituent was introduced to the phenyl ring (**4**  $\text{R}^1 = p\text{-MeOC}_6\text{H}_4$ ) (Table 2; entry 10). Repeating this coupling with *epi*-**4** ( $\text{R}^1 = p\text{-MeOC}_6\text{H}_4$ ), removal of the auxiliary and transformation to the oxazolidinone followed by deprotection of the primary alcohol gave a high yield of (–)-cytoxazone (**8**, Scheme 3).<sup>19,20</sup> Cytoxazone is a cytokine modulator, which has been isolated from *Streptomyces* sp. and has been shown to inhibit Th2 cell's signalling pathway.<sup>21,22</sup>

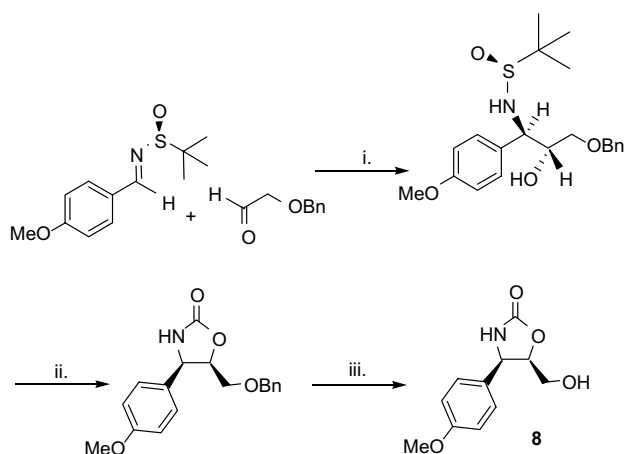
**Table 2.** Cross-coupling of aldehydes with an imine (*R*)-*N*-*tert*-butanesulfinyl auxiliary shown in Scheme 2

Entry	$\text{R}^1$	$\text{R}^2$	Reaction time (h)	Dr <sup>a</sup> crude	Dr <sup>a</sup> after separation	Yield (%)
1	Ph	Et	4	8:1	9:1	90
2	Ph	<sup>i</sup> Pr	3	14:1	>25:1	91
3	Ph	<sup>t</sup> Bu	20	>25:1	>25:1	31
4	Ph	<sup>n</sup> Bu	3	6:1	10:1	86
5	Ph	<sup>n</sup> Bu	3	8:1	9:1	69
6	Ph	<sup>c</sup> C <sub>6</sub> H <sub>11</sub>	3	7:1	>25:1	88
7	Ph	BnOCH <sub>2</sub>	3	—	10:1	70
8	<i>p</i> -ClC <sub>6</sub> H <sub>4</sub>	<sup>s</sup> Bu	3	5:1	10:1	73
9	<i>p</i> -MeOC <sub>6</sub> H <sub>4</sub>	<sup>s</sup> Bu	4	14:1	>25:1	83
10	<i>p</i> -MeOC <sub>6</sub> H <sub>4</sub>	BnOCH <sub>2</sub>	8	15:1	>25:1	85
11	<sup>i</sup> Pr	<sup>i</sup> Pr	10	8:1	>25:1	51

<sup>a</sup> Diastereomeric ratio determined by <sup>1</sup>H NMR.

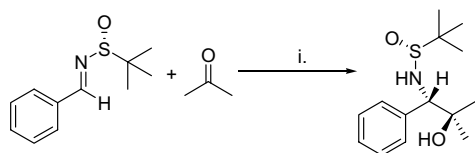


**Figure 1.** The ORTEP image of  $\beta$ -amino alcohol **6** ( $R^1 = p\text{-MeOC}_6\text{H}_4$ ,  $R^2 = ^s\text{Bu}$ ).



**Scheme 3.** The synthesis of (–)-cytoxazone. Reagents and conditions: i.  $\text{SmI}_2$ ,  $^t\text{BuOH}$ , THF,  $-78^\circ\text{C}$  (83%); ii. (a)  $\text{HCl}$ ,  $\text{MeOH}$ ,  $25^\circ\text{C}$  (b) Triphosgene,  $\text{Et}_3\text{N}$ , DCM,  $25^\circ\text{C}$  (85%); iii.  $\text{Pd}(\text{OH})_2/\text{C}$ ,  $\text{H}_2(\text{g})$ ,  $\text{MeOH}$  (86%).

Application of the chiral *N*-*tert*-butanesulfinyl imine auxiliary was further diversified with its coupling of ketones (Scheme 4). Acetone was crossed coupled with the imine (Scheme 4) initially in poor yield but variation of



**Scheme 4.** The synthesis of a  $\beta$ -amino alcohol by the cross-coupling of ketones with an imine (*R*)-*N*-*tert*-butanesulfinyl auxiliary. Reagents and conditions: i.  $\text{SmI}_2$ ,  $^t\text{BuOH}$ , THF,  $-78^\circ\text{C}$ .

**Table 3.** Cross-coupling of ketones with an imine *N*-*tert*-butanesulfinyl auxiliary shown in Scheme 4

Entry	Additive	Reaction time (h)	Yield (%)
1	—	10	33
2	$\text{Et}_3\text{N}$	2	53
3	TMEDA	0.5	56

an amine additive resulted in a faster reaction that gave the  $\beta$ -amino alcohol as a single diastereomer in moderate yield (Table 3).

The very efficient synthesis of  $\beta$ -amino alcohols by the highly stereoselective cross-coupling of *N*-*tert*-butanesulfinyl imine and aldehyde (complementing Uemura's methodology<sup>12,11</sup>) offers a broad scope towards numerous natural product/drug targets, which our group is in the process of exploiting. The imine ketone cross-coupling is being studied further with unsymmetric ketones.

## Acknowledgements

Financial support from the Department of Chemistry and the Research Allocation Committee, University of New Mexico is gratefully acknowledged. We would like to thank Eileen Duesler for assistance with X-ray analysis.

## Supplementary data

Supplementary data associated with this article can be found, in the online version, at [doi:10.1016/j.tetlet.2005.09.002](https://doi.org/10.1016/j.tetlet.2005.09.002).

## References and notes

- Regan, A. C. *J. Chem. Soc., Perkin Trans. 1* **1999**, 357–373.
- Mason, G.; Vallée, S. *Angew. Chem., Int. Ed.* **2002**, *41*, 1772–1775.
- Bravo, P.; Crucianelli, M.; Ono, T.; Zanda, M. *J. Fluorine Chem.* **1999**, *97*, 27–49.
- Kobayashi, J.; Ishibashi, M. *Heterocycles* **1996**, *42*, 943–970.
- Lee, H.-S.; Kang, S. H. *Synth. Lett.* **2004**, 1673–1685.
- Bergmeier, S. C. *Tetrahedron* **2000**, *56*, 2561–2576.
- Nilov, D.; Reiser, O. *Adv. Synth. Catal.* **2002**, *344*, 1169–1173.
- Periasamy, M. *Aldrichim. Acta* **2002**, *35*, 89–101.
- Reetz, M. T. *Chem. Rev.* **1999**, *99*, 1121–1162.
- Tanaka, Y.; Taniguchi, N.; Uemura, M. *J. Org. Chem.* **2002**, *67*, 9227–9237.
- Taniguchi, N.; Uemura, M. *J. Am. Chem. Soc.* **2000**, *122*, 8301–8302.
- Tanaka, Y.; Taniguchi, N.; Uemura, M. *Org. Lett.* **2002**, *4*, 835.
- Zhong, Y.-W.; Dong, Y.-Z.; Fang, K.; Izumi, K.; Xu, M.-H.; Lin, G.-Q. *J. Am. Chem. Soc.* **2005**, *127*, 11956–11957.
- Friestad, G. K.; Draghici, C.; Soukri, M.; Qin, J. *J. Org. Chem.* **2005**, *70*, 6330–6338.

15. Ellman, J. A.; Owens, T. D.; Tang, T. P. *Acc. Chem. Res.* **2002**, 35, 984–995.
16. Zhong, Y.-W.; Izumi, K.; Xu, M.-H.; Lin, G.-Q. *Org. Lett.* **2004**, 6, 4747–4750.
17. Zhong, Y.-W.; Xu, M.-H.; Lin, G.-Q. *Org. Lett.* **2004**, 6, 3953–3956.
18. Andersson, P. G.; Schink, H. E.; Österlund, K. *J. Org. Chem.* **1998**, 63, 8067–8070.
19. Kim, J. D.; Kim, I. S.; Jin, C. H.; Zee, O. P.; Joung, H. J. *Org. Lett.* **2005**, 7, 4025–4028.
20. Miyata, O.; Koizumi, T.; Asai, H.; Iba, R.; Naito, T. *Tetrahedron* **2004**, 60, 3893–3914.
21. Kakeya, H.; Morishita, M.; Kobinata, K.; Osono, M.; Osada, H. *J. Antibiot.* **1998**, 51, 1126–1128.
22. Kakeya, H.; Morishita, M.; Koshino, H.; Morita, T.; Kobayashi, K.; Osada, H. *J. Org. Chem.* **1999**, 64, 1052–1053.